

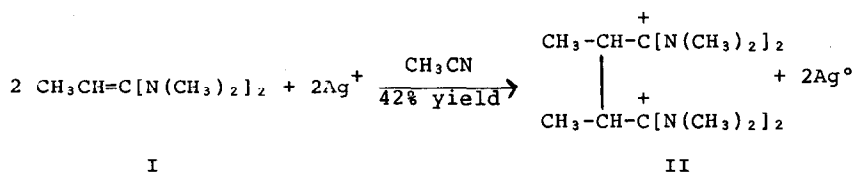
THE OXIDATIVE COUPLING REACTION OF VINYLIDINEBISDIALKYLAMINES

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We wish to report a reaction in which vinylidenebisdiarylamines (1) (ketene-N,N-acetals) are oxidatively coupled to form diimidinium salts. Propenylidenebisdimethylamine, I, for example, reacts exothermically with silver nitrate in acetonitrile solvent to form a mixture of dl and meso-1,1,4,4-tetrakis(dimethylamino)-2,3-dimethylbutane-1,4-diylium nitrate, II. The diylium salt, II, isolated as a hexafluorophosphate, has the following

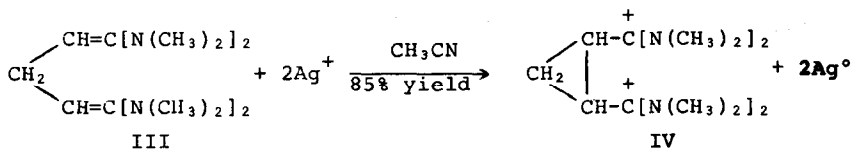


properties: m.p. 245°d, nmr spectrum exhibits a multiplet centered at  $\tau$ 6.35, two singlets at  $\tau$ 6.75 and 6.81 (3:1 dl-meso ratio, not assigned) and two multiplets centered at  $\tau$ 8.54 and 8.66 in the ratio 1:12:3, (acetonitrile- $d_3$ , TMS internal standard).

Anal. Calcd. for  $\text{C}_{14}\text{H}_{32}\text{N}_4\text{P}_2\text{F}_{12}$ : C, 30.8; H, 5.9; N, 10.3.

Found: C, 30.9; H, 6.0; N, 10.3

This method is also applicable to the synthesis of cyclic systems. An example is the oxidation, in good yield, of N,N,N',N',N'',N''',N''',N''''-octamethyl-1,4-pentadiene-1,5-tetramine, III, to a cis and trans mixture of

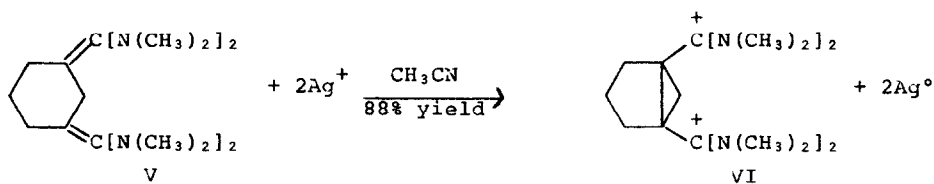


cyclopropylenebis(dimethylaminocarbonium) nitrate, IV. The product, IV, isolated as a hexafluorophosphate, has the following properties: nmr spectrum exhibits two sharp singlets at  $\tau$ 6.68 and 6.80 (cis:trans 2:3) and a poorly defined envelop of peaks between  $\tau$ 7.17 and 8.25 in a ratio 6:1 (acetonitrile- $d_3$ , TMS internal standard),

Anal. Calcd. for  $C_{13}H_{28}N_4P_2F_{12}$ : C, 29.5; H, 5.3; 10.6.

Found: C, 29.7; H, 5.3; N, 10.3.

Furthermore, this method appears suitable for the synthesis of certain bicyclic systems. For example, 1,3-bis[bis(dimethylamino)methylene]cyclohexane, V, is converted to bicyclo[3.1.0]cyclohexan-1,5-ylenebis(dimethylaminocarbonium) nitrate, VI. The diamidinium salt, VI, isolated as a hexafluorophosphate, has the following properties: nmr spectrum exhibits a



singlet at  $\tau$ 6.74 and a complex multiplet between  $\tau$ 7.25 and 8.06 in a ratio 3:1 (acetonitrile- $d_3$ , TMS internal standard), Anal. Calcd. for  $C_{16}H_{32}N_4P_2F_{12}$ : C, 33.7; H, 5.7; N, 9.8 Found: C, 33.4; H, 5.4; N, 9.6.

The structural assignments of II, IV (2) and VI (3) were confirmed by hydrolysis to diamides and comparison to authentic diamides.

ESR and cyclic voltammetry suggest the reaction intermediates are radical cations.

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